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Spring 2020

### Direct Arylation Polymerization of Indophenine-Based Monomers

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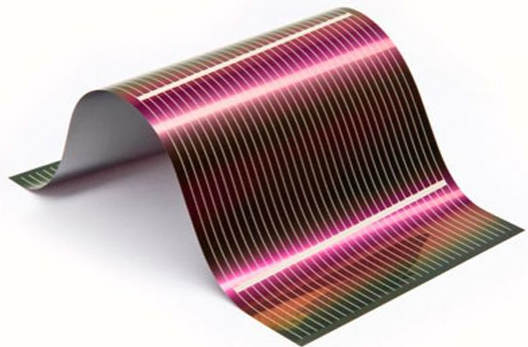
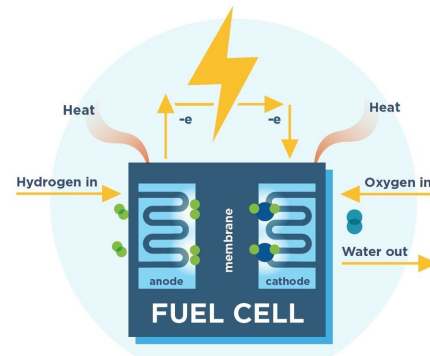
# Direct Arylation Polymerization of Indophenine-Based Monomers

Sarah Severson  
Spring 2020 UROP  
Under the direction of Dr. Ted M. Pappenfus



# Organic redox materials

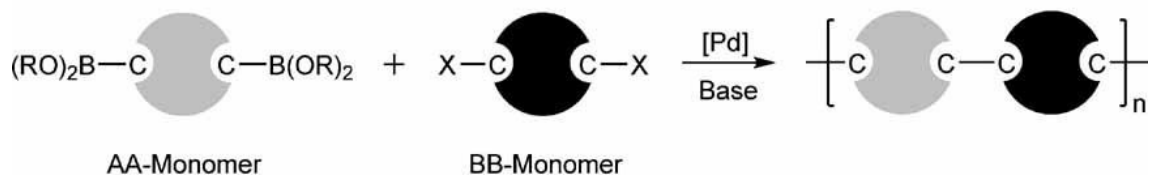
- Low-cost carbon materials
- Unique electronic properties
- High tunability/flexibility



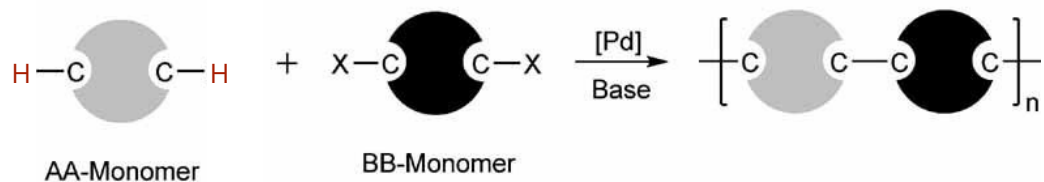


# Polymerization methods

Step-growth polymerization via transition metal-catalyzed coupling (example: Stille)

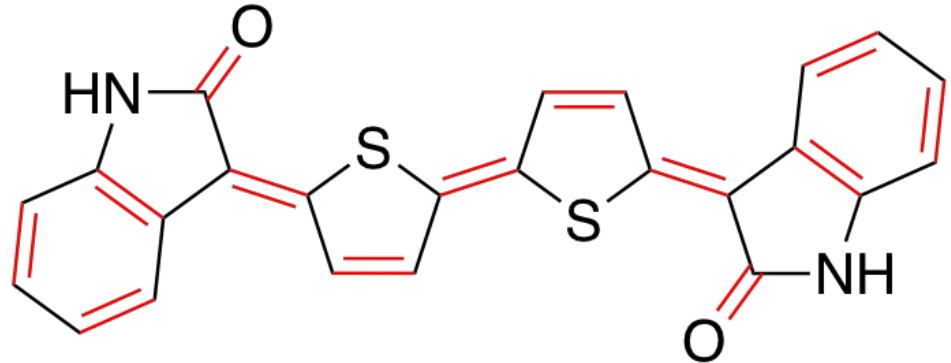


Step-growth polymerization via direct arylation polymerization (DArP)



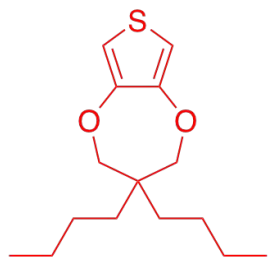
# Indophenine as a monomer for polymerization

- History:
  - **1879:** indophenine was discovered by Alfred Baeyer
  - **1882:** indophenine is made from thiophene (Meyer et al.)
  - **1924:** correct structure identified (Heller et al.)
- Indophenine shows promise for electronic applications due to its...
  - Quinoidal structure
  - Conjugated character
  - Synthetic modifiability



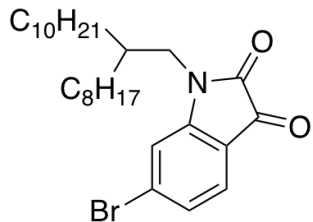
# Indophenine-based monomer

- Modified with ProDOT substituents to enhance solubility and eliminate isomerism (Pappenfus et al.)
- Facile synthesis with acid catalyst (Cava et al.)
- Easily polymerized via Stille coupling

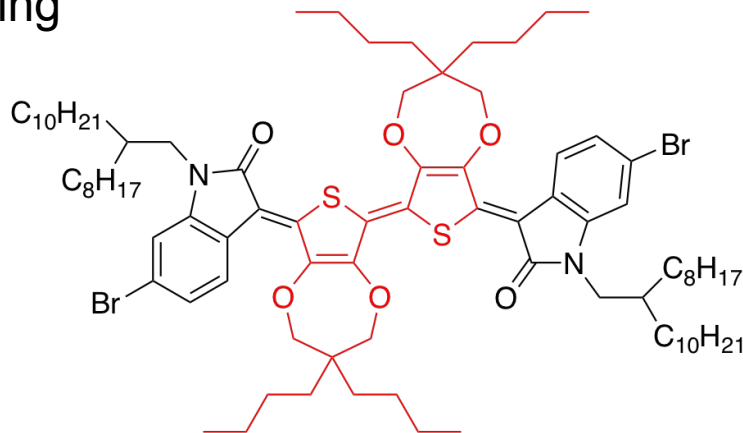
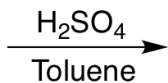


ProDOT

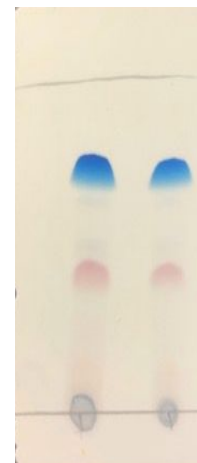
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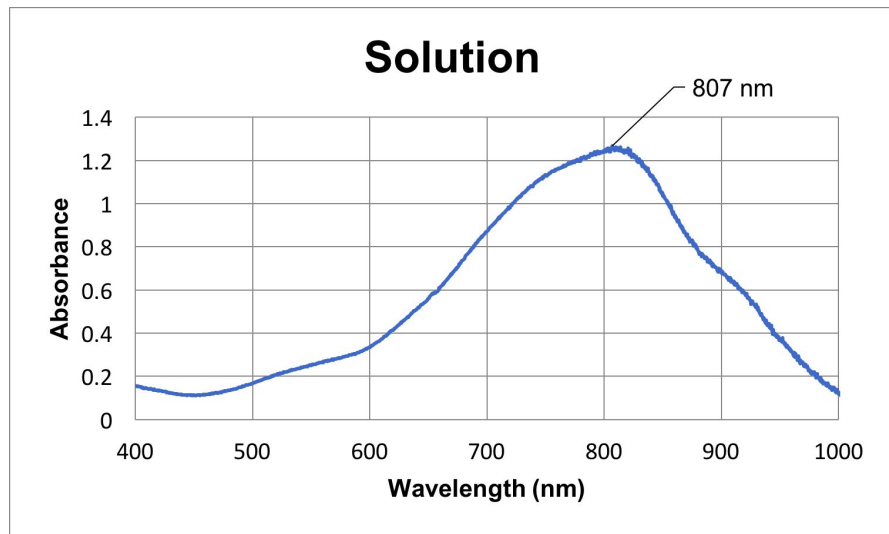
Alkylated-Isatin



Monomer  
32% yield



# Stille copolymer

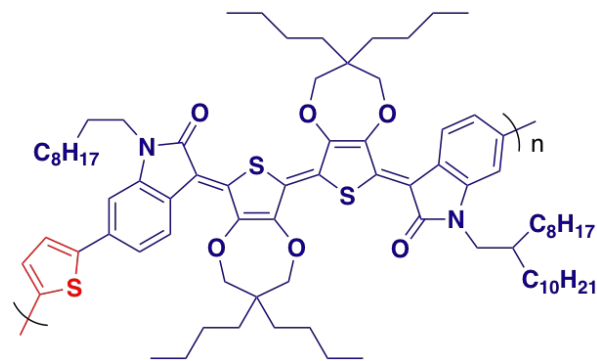


$\lambda_{\text{max}}$  (solution) = 807 nm

$\lambda_{\text{max}}$  (film) = 771 nm

Bandgap = 1.2 eV

<b>Mw</b>	7000 g/mol
<b>Mn</b>	5222
<b>PDI</b>	1.34

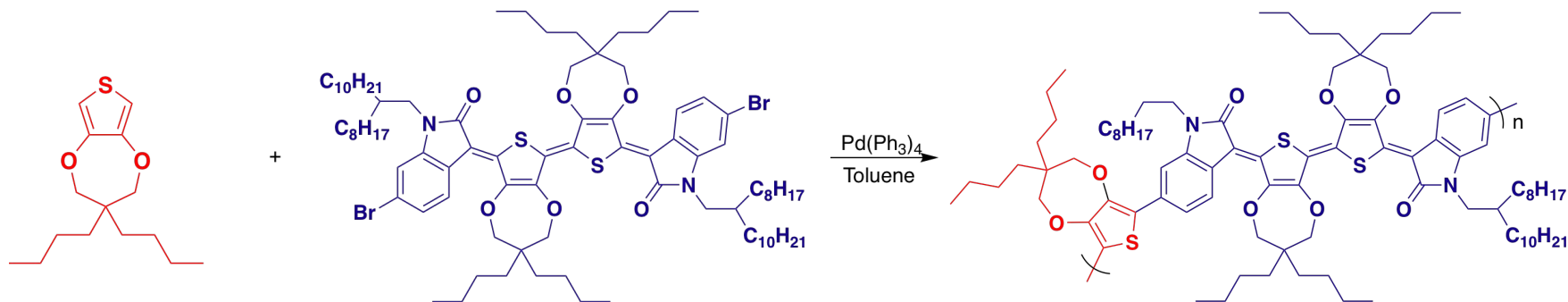


Stille Copolymer  
95% yield



# A greener route: DArP

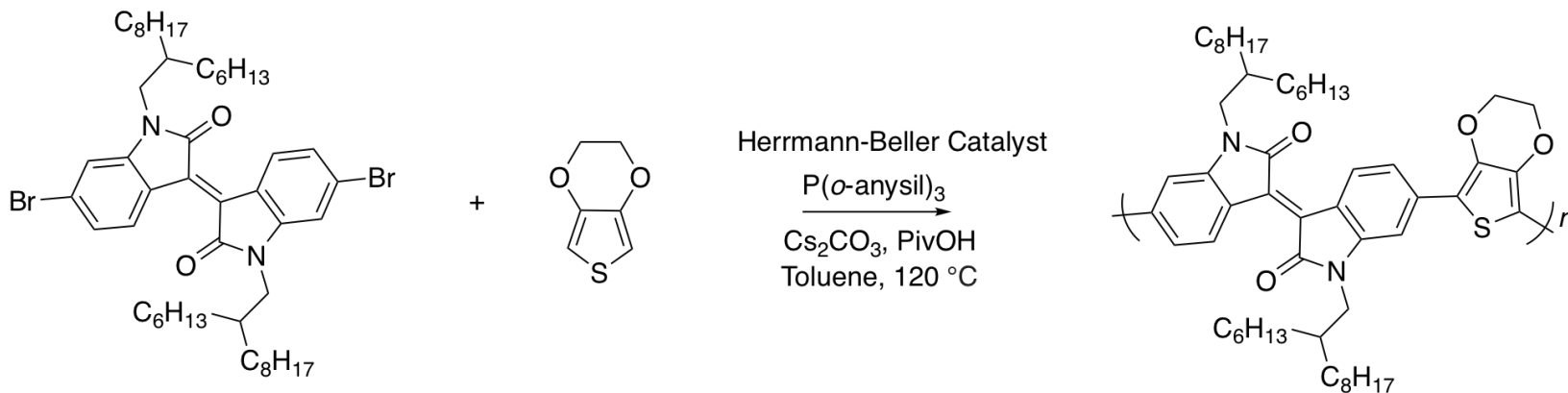
- Requires the functionalization of only *one* coupling site (halide)
- Other coupling site is a C-H bond that is “activated”
- DArP with indophenines has not been previously reported



Preparation of indophenine-based polymer using DArP

# Related DArP polymers

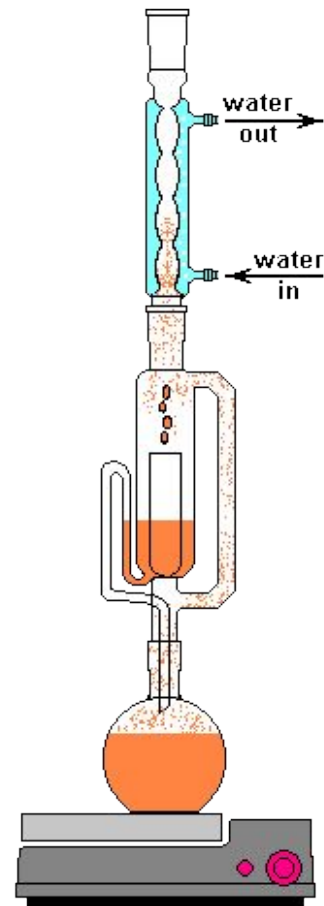
- Grenier et al. (2015) synthesized a similar polymer via DArP with:
  - 95% yield,  $M_w = 210,000$  g/mol, PDI = 2.31



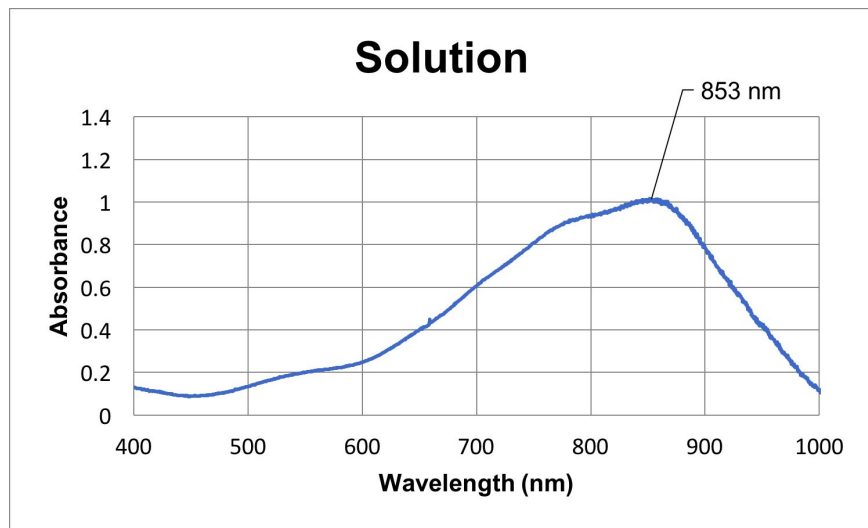
# DArP polymer: purification

Methanol  
Acetone  
Hexanes (33%)  
Chloroform (62%)  
**Combined yield of 95%**

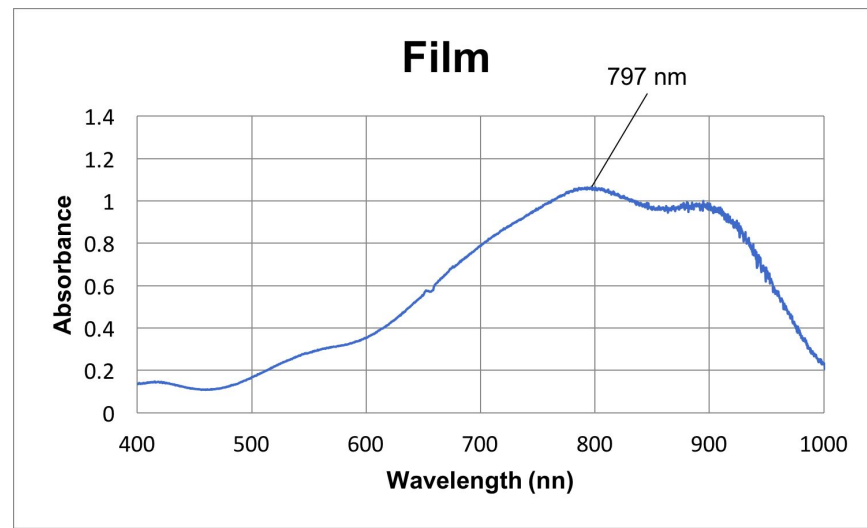
<b>Mw</b>	6047
<b>Mn</b>	5287
<b>PDI</b>	1.13



# DArP polymer: UV-Vis

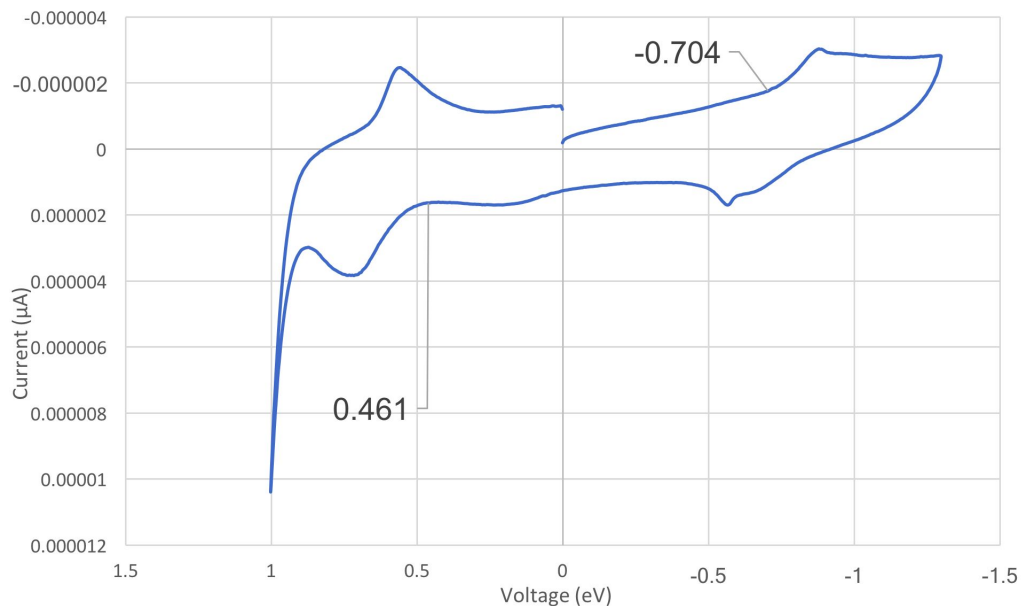
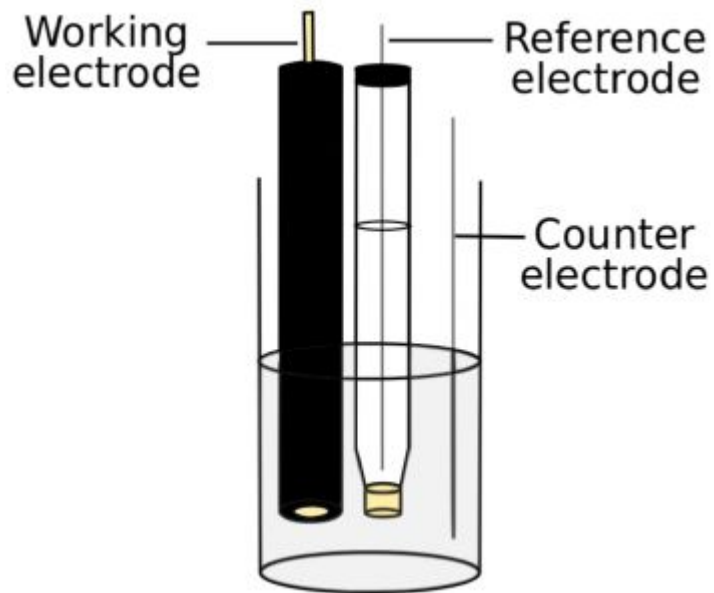


$\lambda_{\text{max}} = 853 \text{ nm}$



$\lambda_{\text{max}} = 797 \text{ nm}$   
*Band gap = 1.22 eV*

# Cyclic voltammetry of DArP polymer



***Electronic Band Gap = 1.17 eV***

# DArP versus Stille Coupling

	Stille polymer	DArP polymer
Yield	95%	95%
Solution $\lambda_{\text{max}}$	807 nm	853 nm
Solution Bandgap	1.2 eV	1.2 eV
Mw	7000 g/mol	6047 g/mol
Mn	5222 g/mol	5287 g/mol
PDI	1.34	1.13

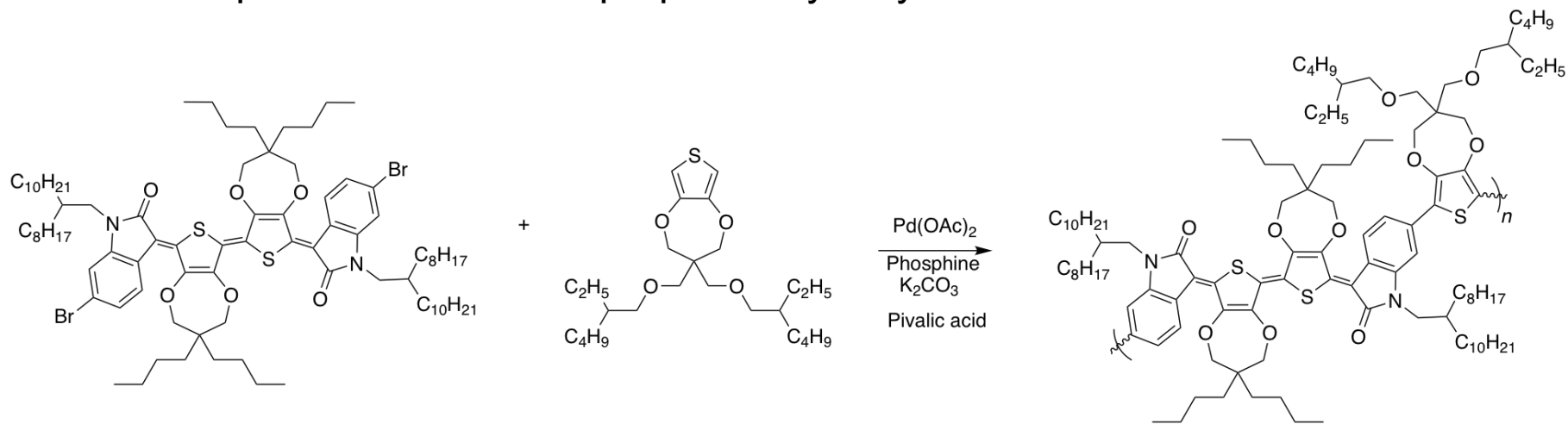
***Direct arylation polymerization is a viable route to produce indophenine polymers with desirable electronic properties.***

# Optimization

- Improving molecular weight by:
  - Enhancing solubility to prevent premature polymerization termination
  - Using already-synthesized alternative monomers with improved solubility
  - Designing new monomers with enhanced solubility
- Increasing sustainability by using alternative solvents
- Exploring new catalytic systems

# Enhancing solubility: thiophene monomer

- Replace ProDOT monomer with a more soluble thiophene monomer
- Used optimized conditions proposed by Meyers et al.

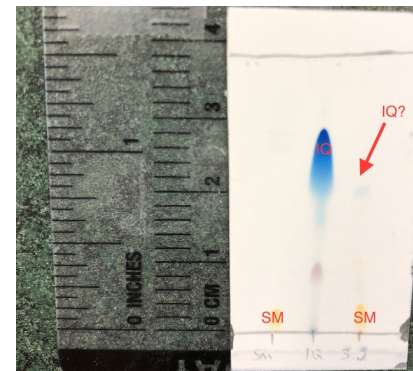
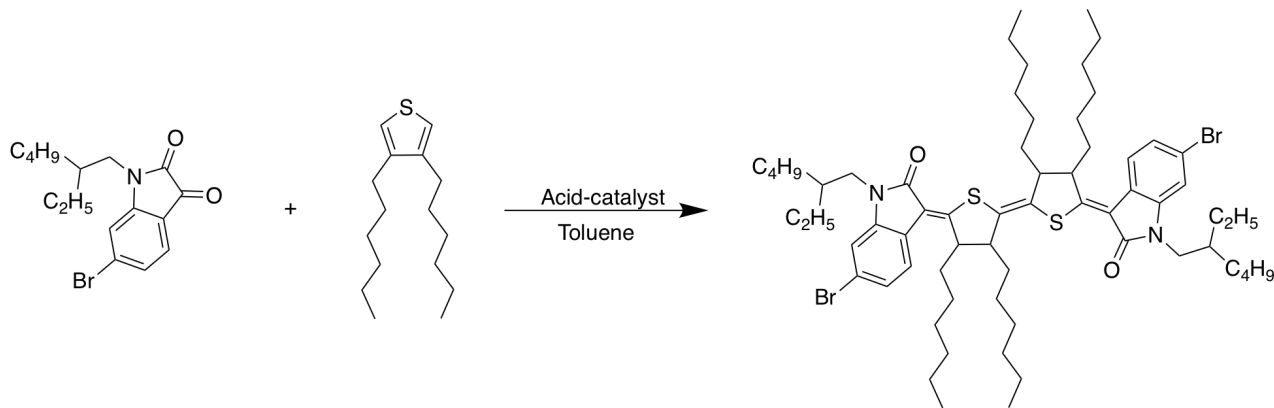


***No visible color change occurred and TLC indicated formation of no product.***



# Enhancing solubility: indophenine monomer

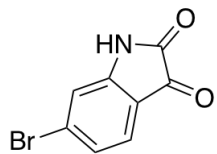
- Design a new indophenine monomer with hexyl substituents rather than ProDOT substituents.



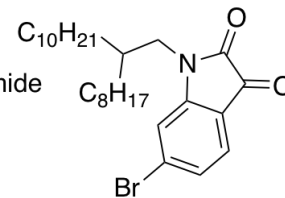
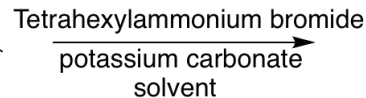
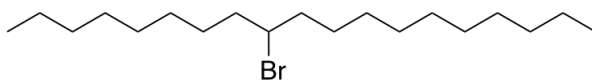
***No visible color change occurred and TLC indicated formation of almost no product. Steric and/or electronic effects likely make the reaction less favorable.***

# Enhancing sustainability: solvent

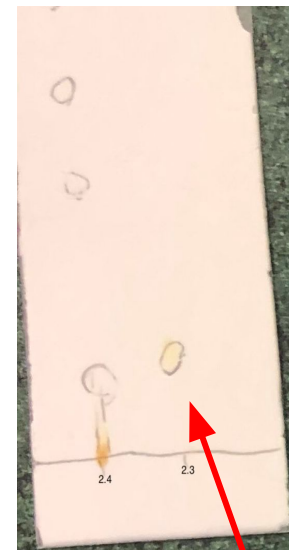
- A key starting material is an alkylated isatin.
- The conventional reaction uses DMF as a solvent, which...
  - Complicates purification
  - Reduces sustainability
- Attempt same reaction with acetone as solvent



+



***In DMF, the reaction formed the desired product in 60% yield. In acetone, no product formed.***



**Only starting material in acetone**

# Conclusions and Future Work

- Direct arylation polymerization is a sustainable and effective route toward indophenine polymers with desirable electronic properties
- Further optimization needed to:
  - Increase the molecular weight of indophenine polymers
  - Enhance the solubility of indophenine monomers and polymers
  - Improve the sustainability of the synthesis of starting materials
- Future work includes:
  - Theoretical calculations to understand energy barriers of failed reactions
  - Continued exploration of more soluble indophenine monomers
  - Continued testing of different catalytic systems

# Acknowledgements

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- UROP facilitators

# References

1. Baeyer, A. Untersuchungen über die Gruppe des Indigblaus. *Ber. Dtsch. Chem. Ges.* **1879**, *12*, 1309–1319.
2. Meyer, V. *Ber. Dtsch. Chem. Ges.* **1882**, *15*, 2893–2894
3. Heller, G. Zur Konstitution des Indophenins. *Angew. Chem.* **1924**, *37*, 1017–103.
4. Tormos, G. V.; Belmore, K. A.; Cava, M. P. The Indophenine Reaction Revisited: Properties of a Soluble Dialkyl Derivative., *J. Am. Chem. Soc.* **1993**, *115*, 11512–11515.
5. Pappenfus, T. M.; Helmin, A. J.; Wilcox, W. D.; Severson, S. M.; Janzen, D. E. ProDOT-Assisted Isomerically Pure Indophenines. *J. Org. Chem.* **2019**, *84*, 11253-11257.
6. Grenier, F.; Aich, B. R.; Lai, Y. Y.; Guerette, M.; Holmes, A. B.; Tao, Ye.; Wong, W. W. H.; Leclerc, M. *Chem. Mater.* **2015**, *27*, 2137-2143.
7. Ludwiczak, M.; Majchrazak, M.; Bayda, M.; Marciniak, Br.; Kubicki, M.; Marciniac, Bo. *J. Organomet. Chem.* **2014**, *750*, 150-161.